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# **CPO Pool Math Workbook**

Math Calculation Formulas Disinfectants Pool Volume Turnover & Flow Rates Water Balance Chemical Adjustment Breakpoint Chlorination

# **MATH CALCULATION FORMULAS**

Amount Conversions	
	Ounces ÷ 16 = Pounds
Fluid Ounces to Gallons =	Fluid Ounces ÷ 128 = Gallons
Distance Conversions	
Yards to Feet =	Yards x 3 = Feet
Meters to Feet =	Meters x 3.28 = Feet
Surface Areas	*Radius = Diameter ÷ 2
Rectangle or Square =	Length x Width = Square Feet
<b>.</b> .	3.14 x Radius x Radius = Square Feet
Pool/Spa Volume	*Average Depth = (Shallow + Deep) ÷ 2
	Length x Width x Average Depth x 7.5 = Gallons
	3.14 x Radius x Radius x Average Depth x 7.5 = Gallons
Gircle -	5.14 x Radius x Radius x Riverage Depth x 7.5 – Ganons
Turnover Rate	
	Pool Volume ÷ Flow Rate ÷ 60 = Hours
nours –	roor volume + riow Rate + 00 - nours
Flow Rate	
Gallons Per Minute (GPM) =	Pool Volume ÷ Turnover Rate ÷ 60 = GPM
Filter Surface Area	
Square Feet =	Flow Rate ÷ Filtering Rate = Square Feet
Heater Sizing	
British Thermal Units (BTU) =	Pool Volume x 8.33 x Temperature Adjustment = BTU
Spa Water Replacement Frequency	
Replacement Internvals (Days) =	Spa Gallons ÷ 3 ÷ Average Users Per Day = Days

## **CHARACTERISTICS OF DISINFECTANTS**

Disinfectants inactivate or kill the vast majority of microorganisms that can cause disease (pathogens). Pathogens include bacteria, fungi, viruses, and protozoan parasites. In addition, disinfectants are effective at killing algae.

	Sodium Hypochlorite	Calcium Hypochlorite	Lithium Hypochlorite	Gas Chlorine	Trichlor	Dichlor	BCDMH (Bromine)
% Available Chlorine	10-12%	47-78%	35%	100%	90%	56-63%	27%
% Active Strength	10-12%	47-78%	29%	100%	>99%	>99%	95.50%
pH in 1% Solution	9-14	8.5-11	10.8	0	2.8-3.5	6.5-6.8	4.8
pH Effect in Water	Raises	Raises	Raises	Lowers	Lowers	Neutral	Lowers
Physical Appearance	Liquid	Granular, Tabs, Briquet	Granular	Gas	Granular, Tabs	Granular	Granular, Tabs

### **POOL & SPA VOLUME**

Pool Volume\*Average Depth = (Shallow + Deep) ÷ 2Rectangle =Length x Width x Average Depth x 7.5 = Gallons

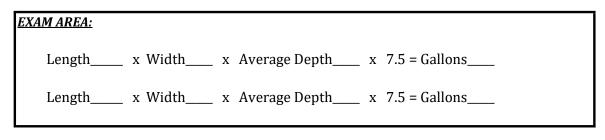
<b>A</b>	120 Ft►	•
65 Ft	Top View	le
V		•

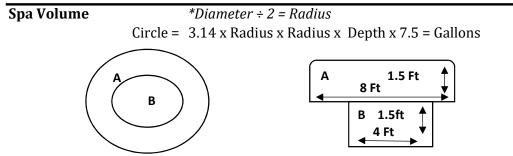


Your pool is 120 ft long by 65 ft wide. The shallow end is 4 ft and the deep end is 6 ft. **What is the volume in gallons?** 

Length\_\_\_\_\_ x Width\_\_\_\_ x Average Depth\_\_\_\_ x 7.5 = Gallons\_\_\_\_ Length <u>120</u> x Width <u>65</u> x Average Depth\_\_\_\_ x 7.5 = Gallons\_\_\_\_ (Shallow <u>4</u> + Deep <u>6</u>) ÷ 2 = \*Average Depth <u>5</u>

Length <u>120</u> x Width <u>65</u> x Average Depth <u>5</u> x 7.5 = Gallons <u>292,500</u>





Your spa diameter is 8 feet, your seating depth is 1.5 feet. The footwell diameter is 4 feet and has a depth of 1.5 ft.

What is the spa volume?

#### TOTAL SPA VOLUME = A VOLUME + B VOLUME

- A 3.14 x Radius \_\_\_\_ x Radius \_\_\_ x Depth \_\_\_ x 7.5 = Gallons \_\_\_\_ 3.14 x Radius <u>4</u> x Radius <u>4</u> x Depth <u>1.5</u> x 7.5 = Gallons <u>565.2</u>
- B
   3.14 x Radius \_\_\_\_ x Radius \_\_\_ x Depth \_\_\_ x 7.5 = Gallons \_\_\_\_

   3.14 x Radius \_2 x Radius \_2 x Depth \_\_\_\_ x 7.5 = Gallons \_\_\_\_\_

TOTAL SPA VOLUME\_\_\_\_ = A VOLUME\_\_\_\_ + B VOLUME\_\_\_\_

TOTAL SPA VOLUME <u>94.2 OR 94 Gallons</u> = A VOLUME <u>75.36</u> + B VOLUME <u>18.84</u>

#### EXAM AREA:

A 3.14 x Radius \_\_\_\_ x Radius \_\_\_\_ x Depth \_\_\_\_ x 7.5 = Gallons \_\_\_\_

**B** 3.14 x Radius \_\_\_\_ x Radius \_\_\_\_ x Depth \_\_\_\_ x 7.5 = Gallons \_\_\_\_

TOTAL SPA VOLUME\_\_\_ = A VOLUME\_\_\_ + B VOLUME\_

### TURNOVER RATE (TR) & FLOW RATES (FR)

<u>*Turnover Rate (TR)*</u> - When the total volume of a pool is passed through its filter within a 6hour period, the pool has had one turnover per 6 hours or 4 turnovers per 24 hours. Codes require that the entire volume of pool water be circulated through the filter in a certain time period. Rates for some water park and speciality pools may vary due to unique conditions, and turnover requirements vary by jurisdictional codes.

*<u>Turnover Rate Standards</u>*: The following turnover rates are considered standard in most jurisdictions. Use these values to calculate the minimum required flow rate for your facility:

Swimming Pools - 6 Hours
Spas - 30 Minutes
Wading Pools - 1 to 2 Hours
Waterparks/specialty pools - 3 to 4 Hours
Hospital.Health Club Pools - as low as 3 Hours
Therapy Pools - 2 Hours

<u>Flow Rate (FR)</u> - The flow rate is the number of gallons per minute passing a given point in the ciruculation system. Find the required flow rate in gallons per minute (gpm) with a known turnover rate and pool volume.

*Formula:* Pool Volume (in gallons) ÷ Turnover Time (in hours) ÷60 (min./hour) = Flow Rate (FR in gpm)

#### Relationship Between Turnover Rate and Flow Rate:

Pool volume (in gallons), FR is the Flow Rate of the pump (in gpm), 60 is the conversion for minutes per hour, and TR is the Turnover Rate

#### **Flow Rate**

Pool Volume ÷ TR (in hours) ÷ 60 min = FR in GPM Example: In a 200,000 gallon pool at a 6 hour turnover rate, what is your flow rate?

> Pool Volume \_\_\_\_\_  $\div$  TR \_\_\_\_\_ (in hours)  $\div$  60 min = FR \_\_\_\_\_ in GPM Pool Volume 200,000  $\div$  TR 6 (in hours)  $\div$  60 min = FR 555.56 in GPM

#### **Turnover Time**

Pool Volume ÷ FR ÷ 60 min/hr = TR in Hours Example: In a 185,000 gallon pool with a FR of 550 gpm has a turnover rate of?

> Pool Volume \_\_\_\_\_  $\div$  FR \_\_\_\_\_  $\div$  60 min/hr = TR \_\_\_\_\_ in Hours Pool Volume <u>185,000</u>  $\div$  FR <u>550</u>  $\div$  60 min/hr = TR <u>5.6</u> in Hours

#### EXAM AREA:

**FR Formula** Pool Volume \_\_\_\_\_ ÷ TR \_\_\_\_\_ (in hours) ÷ 60 min = FR \_\_\_\_\_ in GPM

**TR Formula** Pool Volume \_\_\_\_\_ ÷ FR \_\_\_\_\_ ÷ 60 min/hr = TR\_\_\_\_\_ in Hours

# **WATER BALANCE - SATURATION INDEX**

SI =	pH +	Tf +	Cf +	Af -	TDSf
Saturation Index	pH as Tested	Temperature Factor	Calcium Factor	Alkalinity Factor	TDS Factor

Example:

You are the operator of 100,000 gallon pool. The water tests give you the following readings: Total alkalinity is 75 ppm. pH 7.4 Calcium hardness 125 ppm. Temperature is 84. Total dissolved solids is 500 ppm. What is your saturation index?

Temperature		<b>Calcium Hardness</b> (Expressed as CaCO3)		Total Carbonate Alkalinity	
F	Tf	ppm	Cf	ppm	Af
32	0.0	25	1.0	25	1.4
37	0.1	50	1.3	50	1.7
46	0.2	75	1.5	75	1.9
53	0.3	100	1.6	100	2.0
60	0.4	125	1.7	125	2.1
66	0.5	150	1.8	150	2.2
76	0.6	200	1.9	200	2.3
84	0.7	250	2.0	250	2.4
94	0.8	300	2.1	300	2.5
105	0.9	400	2.2	400	2.6
		800	2.5	800	2.9

#### Saturation Index Factors

	Total D	issolved Solid Fac	tors
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TDS	Factor
≤ 800	12.1
801 - 1,500	12.2
1,501 - 2,900	12.3
2900 - 5,500	12.4
> 5,500	12.5

	Value	Factor
рН	7.4	+7.4
Temperature	84	+0.7
Calcium Hardness	125	+1.7
Carbonate Alkalinity	75	+1.9 🔻
Sub-Total		11.7
Total Dissolved Solids	500	-12.1
Saturation Index		-0.4

EXAM AREA:		Value	Factor
	рН		
	Temperature		
	Calcium Hardness		
	Carbonate Alkalinity		
	Sub-Total		
	Total Dissolved Solids		
	Saturation Index		

# **CHEMICAL ADJUSTMENTS**

Pool Volume _	÷ 10,000 x ppm change	x Chemical Us	ed ÷ Essential Conversion	=
Б	Increase Chlorine	1 ppm	Increase Total Alkalinity	1 ppm
	Chlorine Gas	1.3 oz	Sodium Bicarbonate	2.24 oz
	Calcium Hypochlorite (67%)*	2 oz	Sodium Caronate	1.4 oz
	Sodium Hypochlorite (12%)	10.7 fl oz	Sodium Sesquicarbonate	2 oz
	Lithium Hypochlorite		Sourain Sesquicar bollate	2 UZ
	••	3.8 oz	Deerse wee Tetal Albakista	1
	Dichlor (62%)	2.1 oz	Decrease Total Alkalinity	1 ppm
	Dichlor (56%)	2.4 oz	Muriatic Acid (31.4%)	2.6 fl oz
Ľ	Frichlor	1.5 oz	Sodium Bisulfate	3.36 oz
	Decrease Chlorine	1 ppm	Increase Stabilizer	1 ppm
5	Sodium Thiosulfate	2.6 oz	Cxyanuric Acid	1.3 oz
S	Sodium Sulfite	2.4 oz		
6	Increase Calcium Hardness	1 ppm		
	Caclium Chloride (100%)	1.44 oz	**Appendix B-2 in Pool & Spa	Operator
	Calcium Chloride (77%)	1.44 02 1.92 oz	Handbook	
Ľ		1.72 02		
Г			onvert to pounds or gallons	
			Ounces ÷ 16 = Pounds	
L	Converting Fluid Ounce	s to Gallons: Total	Fluid Ounces ÷ 128 = Gallons	
<u>1</u>	<u>Example:</u> You operate a 15,000 gallon p	ool with a calcium ha	ardness level of 150 ppm. How much increase this pool to 220 ppm?	n calcium
Pool Volume	Example: You operate a 15,000 gallon p chloride (77%) v ÷ 10,000 x ppm change	ool with a calcium ha would be required to x Chemical Us	ardness level of 150 ppm. How much	=
200l Volume 200l Volume <u>1</u> 	Example: You operate a 15,000 gallon p chloride (77%) v ÷ 10,000 x ppm change 5,000 ÷ 10,000 x ppm change <u>7(</u> Example:	ool with a calcium ha would be required to x Chemical Us <u>)</u> x Chemical Used <u>1.</u>	ardness level of 150 ppm. How much increase this pool to 220 ppm? red ÷ Essential Conversion <u>92</u> ÷ Essential Conversion <u>16</u> = <u>12.6</u> wer the free chlorine from 16 ppm to	= <u>lbs</u>
Pool Volume Pool Volume <u>1</u> <u>I</u> Pool Volume	Example: You operate a 15,000 gallon po- chloride (77%) v ÷ 10,000 x ppm change 5,000 ÷ 10,000 x ppm change 7( Example: How much sodium thiosulfate w ÷ 10,000 x ppm change	ool with a calcium ha would be required to x Chemical Us 2 x Chemical Used <u>1.</u> vould you need to lov 60,000 gallo x Chemical Us	ardness level of 150 ppm. How much increase this pool to 220 ppm? red ÷ Essential Conversion <u>92</u> ÷ Essential Conversion <u>16</u> = <u>12.6</u> wer the free chlorine from 16 ppm to	= 2 ppm in a =
200l Volume 200l Volume <u>1</u> <u>1</u> 200l Volume 200l Volume <u>6</u>	Example:You operate a 15,000 gallon po- chloride (77%) v ÷ 10,000 x ppm change 5,000 ÷ 10,000 x ppm change 7( Example: How much sodium thiosulfate w ÷ 10,000 x ppm change 0,000 ÷ 10,000 x ppm change 14 Example:	ool with a calcium ha would be required to x Chemical Us 2 x Chemical Used <u>1.</u> yould you need to lov 60,000 gallo x Chemical Us <u>k</u> x Chemical Used <u>2.</u>	ardness level of 150 ppm. How much increase this pool to 220 ppm? eed ÷ Essential Conversion 92 ÷ Essential Conversion <u>16 = 12.6</u> ver the free chlorine from 16 ppm to n pool? eed ÷ Essential Conversion <u>6</u> ÷ Essential Conversion <u>16 = 13.65</u> eded to raise the free chlorine from 2	= 2 ppm in a = <u>lbs</u>
Pool Volume Pool Volume <u>1</u> <u>1</u> Pool Volume <u>1</u> Pool Volume	Example:         You operate a 15,000 gallon pechloride (77%) weight of the chloride (77%) weight of the	ool with a calcium ha would be required to x Chemical Us <u>)</u> x Chemical Used <u>1</u> . yould you need to lov 60,000 gallo x Chemical Used <u>2</u> . ypochloite will be ne ppm in a 250,000 x Chemical Use	ardness level of 150 ppm. How much increase this pool to 220 ppm? eed ÷ Essential Conversion 92 ÷ Essential Conversion <u>16 = 12.6</u> ver the free chlorine from 16 ppm to n pool? eed ÷ Essential Conversion <u>6</u> ÷ Essential Conversion <u>16 = 13.65</u> eded to raise the free chlorine from 2	= 2 ppm in a = 2 ppm to 4 =
Pool Volume Pool Volume 1 Pool Volume Pool Volume 6 <u>I</u> Pool Volume	Example:         You operate a 15,000 gallon pechloride (77%) weight of the chloride (77%) weight of the	ool with a calcium ha would be required to x Chemical Us <u>)</u> x Chemical Used <u>1</u> . yould you need to lov 60,000 gallo x Chemical Used <u>2</u> . ypochloite will be ne ppm in a 250,000 x Chemical Use	ardness level of 150 ppm. How much increase this pool to 220 ppm? eed ÷ Essential Conversion 92 ÷ Essential Conversion 16 = 12.6 ver the free chlorine from 16 ppm to n pool? eed ÷ Essential Conversion 6 ÷ Essential Conversion 16 = 13.65 eded to raise the free chlorine from 2 gallon pool? eed ÷ Essential Conversion	= 2 ppm in a = 2 ppm to 4 =
ool Volume ool Volume 1 <u>l</u> ool Volume ool Volume 6 <u>l</u> ool Volume 2 ool Volume 2	Example:	ool with a calcium ha would be required to x Chemical Used <u>1.</u> zould you need to lov 60,000 gallo x Chemical Used <u>2.</u> ypochloite will be ne ppm in a 250,000 x Chemical Used <u>10</u> x Chemical Used <u>10</u>	ardness level of 150 ppm. How much increase this pool to 220 ppm? eed ÷ Essential Conversion 92 ÷ Essential Conversion 16 = 12.6 wer the free chlorine from 16 ppm to n pool? eed ÷ Essential Conversion 6 ÷ Essential Conversion 16 = 13.65 eded to raise the free chlorine from 2 gallon pool? eed ÷ Essential Conversion 0.7 ÷ Essential Conversion 128 = 4.1	= 2 ppm in a = <u>lbs</u> 2 ppm to 4 = 8 gallons
ool Volume ool Volume 1  ool Volume ool Volume 6  <u>l</u> ool Volume 2 <u></u> <b>xam Area:</b> ool Volume	Example:       You operate a 15,000 gallon perchloride (77%) weight of the chloride (77%) we	ool with a calcium ha would be required to x Chemical Used <u>1</u> . vould you need to lov 60,000 gallo x Chemical Used <u>2</u> . ypochloite will be ne ppm in a 250,000 x Chemical Used <u>10</u> x Chemical Used <u>10</u> x Chemical Used <u>10</u>	ardness level of 150 ppm. How much increase this pool to 220 ppm? eed ÷ Essential Conversion 92 ÷ Essential Conversion 16 = 12.6 ver the free chlorine from 16 ppm to n pool? eed ÷ Essential Conversion 6 ÷ Essential Conversion 16 = 13.65 eded to raise the free chlorine from 2 gallon pool? eed ÷ Essential Conversion	= 2 ppm in a = 2 ppm to 4 = 8 gallons =
ool Volume ool Volume 1 // ool Volume ool Volume 6 // / / / / / / / / / / / / / / / / /	Example:         You operate a 15,000 gallon prochloride (77%) w         ÷ 10,000 x ppm change         5,000 ÷ 10,000 x ppm change 70         Example:         How much sodium thiosulfate w         ÷ 10,000 x ppm change         0,000 ÷ 10,000 x ppm change 14         Example:         How many gallons of sodium hy         ÷ 10,000 x ppm change         ± 10,000 x ppm change	ool with a calcium ha would be required to x Chemical Used <u>1</u> . vould you need to lov 60,000 gallo x Chemical Used <u>2</u> . vpochloite will be ne ppm in a 250,000 x Chemical Used <u>10</u> x Chemical Used <u>10</u> x Chemical Used <u>10</u> x Chemical Used <u>10</u>	ardness level of 150 ppm. How much increase this pool to 220 ppm? eed ÷ Essential Conversion 92 ÷ Essential Conversion 16 = 12.6 wer the free chlorine from 16 ppm to n pool? eed ÷ Essential Conversion 6 ÷ Essential Conversion 16 = 13.65 eded to raise the free chlorine from 2 gallon pool? eed ÷ Essential Conversion .7 ÷ Essential Conversion 128 = 4.1	= 2 ppm in a = 2 ppm to 4 = 8 gallons =

## **BREAKPOINT CHLORINATION**

The term breakpoint chlorination is used to describe the process of adding free chlorine (FC) to treat, oxidize or remove combined chlorine. To achieve breakpoint, the free chlorine (FC) in the water must be raised to about 10 times the amount of combined chlorine (CC).

	TC - FC =	CC Combined	
	Total Chloine Free Chloring		
#1	TC FC =	CC	
#2	CC x 10 = Breakp	int Chlorination (BPC)	
#3	BPC FC	= PPM Change	
#4	Pool Volume ÷ 10	000 x PPM Change x Chemical* ÷ Conversion = *(See Appendix B-2 for chemical)	

*Essential Conversions: Always convert to pounds or gallons* Converting Ounces to Pounds: Total Ounces ÷ 16 = Pounds Converting Fluid Ounces to Gallons: Total Fluid Ounces ÷ 128 = Gallons

*Example:* You have a 65,000 gallon pool that has a free chlorine (FC) reading of 1.5 ppm and total chlorine (TC) reading of 3 ppm. How much calcium hypochorite is needed to reach breakpoint?

- #1 TC <u>3</u> FC <u>1.5</u> = CC <u>1.5</u>
- #2 CC  $1.5 \times 10$  = Breakpoint Chlorination (BPC) 15
- #3 BPC <u>15</u> FC <u>1.5</u> = PPM Change <u>13.5</u>
- #4 Pool Volume  $\underline{65,000} \div 10,000 \times PPM$  Change  $\underline{13.5} \times Chemical^* \underline{2} \div Conversion \underline{16} = \underline{10.97 \text{ lbs}}$ \*(See Appendix B-2 for chemical)

Exa	m Area:
#1	TC FC = CC
#2	CC x 10 = Breakpoint Chlorination (BPC)
#3	BPC FC = PPM Change
#4	Pool Volume ÷ 10,000 x PPM Change x Chemical* ÷ Conversion = *(See Appendix B-2 for chemical)